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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
10/009,910	12/12/2001	Makoto Iida	81839.0107	7347

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EXAMINER

SONG, MATTHEW J

ART UNIT PAPER NUMBER

1765

DATE MAILED: 05/09/2003

Please find below and/or attached an Office communication concerning this application or proceeding.

Office Action Summary

Application No.

10/009,910

Applicant(s)

IIDA ET AL.

Examiner

Matthew J Song

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-- The MAILING DATE of this communication appears on the cover sheet with the correspondence address --

Period for Reply

A SHORTENED STATUTORY PERIOD FOR REPLY IS SET TO EXPIRE 3 MONTH(S) FROM THE MAILING DATE OF THIS COMMUNICATION.

- Extensions of time may be available under the provisions of 37 CFR 1.136(a). In no event, however, may a reply be timely filed after SIX (6) MONTHS from the mailing date of this communication.
- If the period for reply specified above is less than thirty (30) days, a reply within the statutory minimum of thirty (30) days will be considered timely.
- If NO period for reply is specified above, the maximum statutory period will apply and will expire SIX (6) MONTHS from the mailing date of this communication.
- Failure to reply within the set or extended period for reply will, by statute, cause the application to become ABANDONED (35 U.S.C. § 133).
- Any reply received by the Office later than three months after the mailing date of this communication, even if timely filed, may reduce any earned patent term adjustment. See 37 CFR 1.704(b).

Status

- 1) ☒ Responsive to communication(s) filed on 25 February 2003.
- 2a) ☒ This action is **FINAL**. 2b) ☐ This action is non-final.
- 3) ☐ Since this application is in condition for allowance except for formal matters, prosecution as to the merits is closed in accordance with the practice under *Ex parte Quayle*, 1935 C.D. 11, 453 O.G. 213.

Disposition of Claims

- 4) ☒ Claim(s) 1-18 is/are pending in the application.
- 4a) Of the above claim(s) _____ is/are withdrawn from consideration.
- 5) ☐ Claim(s) _____ is/are allowed.
- 6) ☒ Claim(s) 1-18 is/are rejected.
- 7) ☐ Claim(s) _____ is/are objected to.
- 8) ☐ Claim(s) _____ are subject to restriction and/or election requirement.

Application Papers

- 9) ☐ The specification is objected to by the Examiner.
- 10) ☐ The drawing(s) filed on _____ is/are: a) ☐ accepted or b) ☐ objected to by the Examiner.
- Applicant may not request that any objection to the drawing(s) be held in abeyance. See 37 CFR 1.85(a).
- 11) ☐ The proposed drawing correction filed on _____ is: a) ☐ approved b) ☐ disapproved by the Examiner.
- If approved, corrected drawings are required in reply to this Office action.
- 12) ☐ The oath or declaration is objected to by the Examiner.

Priority under 35 U.S.C. §§ 119 and 120

- 13) ☐ Acknowledgment is made of a claim for foreign priority under 35 U.S.C. § 119(a)-(d) or (f).
- a) ☐ All b) ☐ Some * c) ☐ None of:
1. ☐ Certified copies of the priority documents have been received.
2. ☐ Certified copies of the priority documents have been received in Application No. _____.
3. ☐ Copies of the certified copies of the priority documents have been received in this National Stage application from the International Bureau (PCT Rule 17.2(a)).
- * See the attached detailed Office action for a list of the certified copies not received.
- 14) ☐ Acknowledgment is made of a claim for domestic priority under 35 U.S.C. § 119(e) (to a provisional application).
- a) ☐ The translation of the foreign language provisional application has been received.
- 15) ☐ Acknowledgment is made of a claim for domestic priority under 35 U.S.C. §§ 120 and/or 121.

Attachment(s)

- 1) ☒ Notice of References Cited (PTO-892)
- 2) ☐ Notice of Draftsperson's Patent Drawing Review (PTO-948)
- 3) ☐ Information Disclosure Statement(s) (PTO-1449) Paper No(s) _____
- 4) ☐ Interview Summary (PTO-413) Paper No(s). _____
- 5) ☐ Notice of Informal Patent Application (PTO-152)
- 6) ☐ Other: _____

DETAILED ACTION

Claim Rejections - 35 USC § 103

1. The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negated by the manner in which the invention was made.

2. Claims 1 and 5 are rejected under 35 U.S.C. 103(a) as being unpatentable over Iida et al (US 5,968,264) in view of Fujikawa (US 6,277,501).

Iida et al discloses a method of forming a silicon wafer with an N region formed over the entire surface by pulling a crystal from a silicon melt in a Czochralski method at a pulling rate, V, ranging between 0.55-0.58 mm/min and a G ranging from 42.0-45.0 °C/cm from the center to the edge of the silicon ingot, this reads on applicant's controlling V/G because V and G are controlled, therefore the ratio is inherently controlled (Example 1 and 2). Iida et al also discloses in order to establish the N region over the entire cross section of a crystal, a highly precisely control must be carried out. Also note that the entire reference has been incorporated into the basis of the rejection.

Iida et al does not disclose the silicon single crystal is pulled while doping with carbon.

In a method of forming a silicon wafer, note entire reference, Fujikawa teaches growing a silicon single crystal while controlling the oxygen concentration in the range of 12×10^{17} - 18×10^{17} atoms/cm³ and controlling the carbon concentration in the range of 0.3×10^{16} - 2.5×10^{16} atoms/cm³ (col 9, ln 1-67), where 2.5×10^{16} atoms/cm³ of carbon approximately corresponds to 0.5 ppma (col 5, ln 1-67). Fujikawa also teaches annealing a wafer, containing specified amounts of

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oxygen and carbon, is annealed at 600-900°C for at least more than 15 minutes to achieve a BMD of over $3 \times 10^8/\text{cm}^3$ (col 11, ln 1-67). It would have been obvious to a person of ordinary skill in the art at the time of the invention to modify Iida et al with Fujikawa to promote precipitation of oxygen, thereby producing an epi-wafer without an expensive EG treatment (col 6, ln 1-67 and col 7, ln 1-67).

Referring to claim 5, the combination of Iida et al and Fujikawa teaches annealing at 600-900°C, overlapping ranges are held to be obvious (MPEP 2144.05).

3. Claims 2, 6, 9-18 are rejected under 35 U.S.C. 103(a) as being unpatentable over Iida et al (5,968,264) in view of Fujikawa (US 6,277,501) as applied to claim 1 above, and further in view of Tamatsuka et al (US 6,162,708).

The combination of Iida et al and Fujikawa teaches all of the limitations of claim 2, as discussed previously in claim 1, except the silicon single crystal is doped with nitrogen.

In a method of forming an epitaxial silicon wafer, note entire reference, Tamatsuka et al teaches a silicon single crystal doped with nitrogen in the range of 1×10^{10} to 5×10^{15} atoms/cm³ and an interstitial oxygen concentration in the single crystal ingot is 18 ppma or less (col 2, ln 1-67). Tamatsuka et al also teaches annealing at 900°C (col 8, ln 1-67). It would have been obvious to a person of ordinary skill in the art at the time of the invention to modify the combination of Iida et al and Fujikawa with Tamatsuka et al because a silicon single crystal wafer produced by doping nitrogen during growth of the silicon crystal ingot has a high gettering capability, growth of grown in defects incorporated can be suppressed and density of oxide precipitates can be increased (col 6, ln 1-67).

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Referring to claim 6, the combination of Iida, Fujikawa and Tamatsuka et al teaches annealing at 600-900°C. Overlapping ranges are held to be obvious.

Referring to claims 11-18, the combination of Iida, Fujikawa and Tamatsuka teaches pulling a silicon single crystal with only a N-region, this reads on applicant's pulled not generate secondary defects, with a carbon concentration of 0.05 ppma, a nitrogen concentration of 1×10^{10} - 1×10^{15} atoms/cm³ and a oxygen concentration of 18 ppma or less and annealing to obtain BMD of over 3×10^8 /cm³. Overlapping ranges are held to be obvious.

4. Claims 3 and 7 are rejected under 35 U.S.C. 103(a) as being unpatentable over Iida et al (US 5,968,264) in view of Fujikawa (US 6,277,501) as applied to claim 1 above, and further in view of Hourai et al (US 5,954,873).

The combination of Iida et al and Fujikawa teaches all of the limitations of claim 3, as discussed previously, except controlling V/G within a range of 0.183-0.177 mm²/K min.

Hourai et al discloses a V/G ratio of 0.183-0.177 mm²/K min (Fig 2), where dislocation clusters form through the entire wafer, this reads applicant's N-region, where wafers are formed from a silicon single crystal ingot manufactured by the Czochralski method with careful control of the pulling rate and temperature gradient permit a crystal to be formed that is free of Oxidation induced stacking fault rings and other defects (Abstract). Hourai et al also teaches V and G are important parameters for controlling the diameter of an OSF ring (col 4, ln 1-67). It would have been obvious to a person of ordinary skill in the art at the time of the invention to modify the combination of Iida et al and Fujikawa with Hourai et al because a larger V/G allows the crystal to be pulled faster, thereby increasing production.

5. Claims 4 and 8 are rejected under 35 U.S.C. 103(a) as being unpatentable over Iida et al (US 5,968,264) in view of Fujikawa (US 6,277,501) and Tamatsuka et al (US 6,162,708) as applied to claim 2 above, and further in view of Hourai et al (US 5,954,873).

The combination of Iida et al, Fujikawa and Tamatsuka et al teaches all of the limitations of claim 4, as discussed previously, except controlling V/G within a range of 0.183-0.177 mm²/K min.

Hourai et al discloses a V/G ratio of 0.183-0.177 mm²/K min (Fig 2), where dislocation clusters form through the entire wafer, this reads applicant's N-region, where wafers are formed from a silicon single crystal ingot manufactured by the Czochralski method with careful control of the pulling rate and temperature gradient permit a crystal to be formed that is free of Oxidation induced stacking fault rings and other defects (Abstract). Hourai et al also teaches V and G are important parameters for controlling the diameter of an OSF ring (col 4, ln 1-67). It would have been obvious to a person of ordinary skill in the art at the time of the invention to modify the combination of Iida et al, Fujikawa and Hourai et al with Hourai et al because a larger V/G allows the crystal to be pulled faster, thereby increasing production.

6. Claims 1, 3, 5, 7 and 9 are rejected under 35 U.S.C. 103(a) as being unpatentable over Hourai et al (US 5,954,873) in view of Fujikawa (US 6,277,501).

Hourai et al discloses a V/G ratio of 0.183-0.177 mm²/K min (Fig 2), where dislocation clusters form through the entire wafer, this reads applicant's N-region, where wafers are formed from a silicon single crystal ingot manufactured by the Czochralski method with careful control

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of the pulling rate and temperature gradient permit a crystal to be formed that is free of Oxidation induced stacking fault rings and other defects (Abstract). Hourai et al also teaches V and G are important parameters for controlling the diameter of an OSF ring (col 4, ln 1-67). Also note the entire reference has been incorporated into the basis of the rejection.

Hourai et al does not disclose the silicon single crystal is pulled while doping with carbon

In a method of forming a silicon wafer, note entire reference, Fujikawa teaches growing a silicon single crystal while controlling the oxygen concentration in the range of 12×10^{17} - 18×10^{17} atoms/cm³ and controlling the carbon concentration in the range of 0.3×10^{16} - 2.5×10^{16} atoms/cm³ (col 9, ln 1-67), where 2.5×10^{16} atoms/cm³ of carbon approximately corresponds to 0.5 ppma (col 5, ln 1-67). Fujikawa also teaches annealing a wafer, containing specified amounts of oxygen and carbon, is annealed at 600-900°C for at least more than 15 minutes to achieve a BMD of over 3×10^8 /cm³ (col 11, ln 1-67). It would have been obvious to a person of ordinary skill in the art at the time of the invention to modify Hourai et al with Fujikawa to promote precipitation of oxygen, thereby producing an epi-wafer without an expensive EG treatment (col 6, ln 1-67 and col 7, ln 1-67).

Referring to claim 3, 5 and 7, the combination of Hourai et al and Fujikawa teaches a carbon concentration of 0.5 ppma and a V/G of 0.183-0.177 mm²/K min and annealing at a temperature of 600-900°C. Overlapping ranges are held to be obvious (MPEP 2144.05).

Referring to claim 9, the combination of Hourai et al and Fujikawa teaches a wafer with dislocation clusters throughout the wafer pulled under a similar V/G condition, as applicant, therefore this reads on applicant's N-region. And a carbon concentration of 0.5 ppma.

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7. Claims 2, 4, 6, 8 and 10-18 are rejected under 35 U.S.C. 103(a) as being unpatentable over Hourai et al (US 5,954,873) in view of Fujikawa (US 6,277,501) as applied to claim 1 above, and further in view of Tamatsuka et al (US 6,162,708).

The combination of Hourai et al and Fujikawa teaches all of the limitations of claim 2, as discussed previously in claim 1, except doping with nitrogen.

In a method of forming an epitaxial silicon wafer, note entire reference, Tamatsuka et al teaches a silicon single crystal doped with nitrogen in the range of 1×10^{10} to 5×10^{15} atoms/cm³ and an interstitial oxygen concentration in the single crystal ingot is 18 ppma or less (col 2, ln 1-67). Tamatsuka et al also teaches annealing at 900°C (col 8, ln 1-67). It would have been obvious to a person of ordinary skill in the art at the time of the invention to modify the combination of Hourai et al and Fujikawa with Tamatsuka et al because a silicon single crystal wafer produced by doping nitrogen during growth of the silicon crystal ingot has a high gettering capability, growth of grown in defects incorporated can be suppressed and density of oxide precipitates can be increased (col 6, ln 1-67).

Referring to claims 6 and 8, the combination of Hourai, Fujikawa and Tamatsuka teaches annealing at 600-900°C, overlapping ranges are held to be obvious (MPEP 2144.05).

Referring to claim 10, the combination of Hourai, Fujikawa and Tamatsuka teaches a nitrogen content of 1×10^{10} - 1×10^{15} number/cm³, overlapping ranges are obvious.

Referring to claims 11-18, the combination of Hourai, Fujikawa and Tamatsuka teaches pulling a silicon single crystal to form only dislocation clusters, this reads on applicant's pulled not generate secondary defects, with a carbon concentration of 0.05 ppma, a nitrogen

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concentration of 1×10^{10} - 1×10^{15} atoms/cm³ and a oxygen concentration of 18 ppma or less and annealing to obtain a BMD of over 3×10^8 /cm³. Overlapping ranges are held to be obvious.

Response to Arguments

8. Applicant's arguments filed 2/25/2003 have been fully considered but they are not persuasive.

In response to applicant's arguments against the references individually (pg 6, first full paragraph), one cannot show nonobviousness by attacking references individually where the rejections are based on combinations of references. See *In re Keller*, 642 F.2d 413, 208 USPQ 871 (CCPA 1981); *In re Merck & Co.*, 800 F.2d 1091, 231 USPQ 375 (Fed. Cir. 1986).

In response to applicant's argument that the references fail to show certain features of applicant's invention, it is noted that the features upon which applicant relies (i.e., promoting precipitation of oxygen), note page 6, lines 6-7, are not recited in the rejected claim(s). Although the claims are interpreted in light of the specification, limitations from the specification are not read into the claims. See *In re Van Geuns*, 988 F.2d 1181, 26 USPQ2d 1057 (Fed. Cir. 1993).

In response to applicant's argument that the examiner's conclusion of obviousness is based upon improper hindsight reasoning (pg 6, lines 10-12), it must be recognized that any judgment on obviousness is in a sense necessarily a reconstruction based upon hindsight reasoning. But so long as it takes into account only knowledge which was within the level of ordinary skill at the time the claimed invention was made, and does not include knowledge gleaned only from the applicant's disclosure, such a reconstruction is proper. See *In re*

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McLaughlin, 443 F.2d 1392, 170 USPQ 209 (CCPA 1971). Furthermore, the motivation to combine the references was taken directly from the prior art.

In response to applicant's argument that the single crystal having the N-region can be pulled faster than a single crystal in the case of not doping with carbon (pg 6, second full paragraph), the fact that applicant has recognized another advantage which would flow naturally from following the suggestion of the prior art cannot be the basis for patentability when the differences would otherwise be obvious. See *Ex parte Obiaya*, 227 USPQ 58, 60 (Bd. Pat. App. & Inter. 1985).

In response to applicant's arguments against the references individually (pg 7, first full paragraph), one cannot show nonobviousness by attacking references individually where the rejections are based on combinations of references. See *In re Keller*, 642 F.2d 413, 208 USPQ 871 (CCPA 1981); *In re Merck & Co.*, 800 F.2d 1091, 231 USPQ 375 (Fed. Cir. 1986).

Applicant's arguments, page 6, lines 2-9, fail to comply with 37 CFR 1.111(b) because they amount to a general allegation that the claims define a patentable invention without specifically pointing out how the language of the claims patentably distinguishes them from the references. Applicants merely state a person of ordinary skill in the art would not be lead to the present invention from such combination of reference without providing evidence.

In response to applicant's argument that the range of pulling rate possible to obtain an N-region is expanded when nitrogen is doped (pg 8, First paragraph), the fact that applicant has recognized another advantage which would flow naturally from following the suggestion of the prior art cannot be the basis for patentability when the differences would otherwise be obvious. See *Ex parte Obiaya*, 227 USPQ 58, 60 (Bd. Pat. App. & Inter. 1985).

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In response to applicant's argument that the references fail to show certain features of applicant's invention, it is noted that the features upon which applicant relies (i.e., pulling rate can be pulled faster and the acceleration effect of oxygen precipitation), note page 9, lines 1-5, are not recited in the rejected claim(s). Although the claims are interpreted in light of the specification, limitations from the specification are not read into the claims. See *In re Van Geuns*, 988 F.2d 1181, 26 USPQ2d 1057 (Fed. Cir. 1993).

In response to applicant's arguments against the references individually, one cannot show nonobviousness by attacking references individually where the rejections are based on combinations of references. See *In re Keller*, 642 F.2d 413, 208 USPQ 871 (CCPA 1981); *In re Merck & Co.*, 800 F.2d 1091, 231 USPQ 375 (Fed. Cir. 1986). The combination of Hourai et al and Fujikawa et al teach the claimed invention.

Applicant's arguments, page 10, lines 2-9, fail to comply with 37 CFR 1.111(b) because they amount to a general allegation that the claims define a patentable invention without specifically pointing out how the language of the claims patentably distinguishes them from the references. Applicants merely state a person of ordinary skill in the art would not be lead to the present invention from such combination of reference without providing evidence.

Conclusion

9. The prior art made of record and not relied upon is considered pertinent to applicant's disclosure.

Minami et al (US 6,517,632) teaches doping a silicon using a Czochralski method with carbon and nitrogen, note abstract.

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Togashi et al (US 2003/0068502) claims growing a perfect silicon single crystal ingot and doping with nitrogen, note claims 4-5.

Tamatsuka et al (US 6,478,883) teaches doping a silicon single crystal growing by the Czochralski method with nitrogen (Abstract).

Tamatsuka (US 6,224,668) teaches doping a silicon single crystal growing by the Czochralski method with nitrogen (Abstract).

Iida et al (US 2003/0015131) is a pending application with a common assignee and claimed maintaining V/G ratio and doping with nitrogen, note claims 8-22.

Pyi (US 2003/0079677) teaches doping a silicon ingot with carbon (Abstract).

Kirscht et al (US 6,491,752) teaches co-doping a silicon ingot with carbon (Abstract).

10. Applicant's amendment necessitated the new ground(s) of rejection presented in this Office action. Accordingly, **THIS ACTION IS MADE FINAL**. See MPEP § 706.07(a). Applicant is reminded of the extension of time policy as set forth in 37 CFR 1.136(a).

A shortened statutory period for reply to this final action is set to expire **THREE MONTHS** from the mailing date of this action. In the event a first reply is filed within **TWO MONTHS** of the mailing date of this final action and the advisory action is not mailed until after the end of the **THREE-MONTH** shortened statutory period, then the shortened statutory period will expire on the date the advisory action is mailed, and any extension fee pursuant to 37 CFR 1.136(a) will be calculated from the mailing date of the advisory action. In no event, however, will the statutory period for reply expire later than **SIX MONTHS** from the date of this final action.

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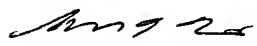
11. Any inquiry concerning this communication or earlier communications from the examiner should be directed to Matthew J Song whose telephone number is 703-305-4953. The examiner can normally be reached on M-F 9:00-5:00.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Benjamin L Utech can be reached on 703-308-3868. The fax phone numbers for the organization where this application or proceeding is assigned are 703-872-9310 for regular communications and 703-872-9311 for After Final communications.

Any inquiry of a general nature or relating to the status of this application or proceeding should be directed to the receptionist whose telephone number is 703-308-0661.

Matthew J Song
Examiner
Art Unit 1765

MJS
May 5, 2003


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